

Exploration Guide Covalent Bonds

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Student Exploration: Covalent Bonds. Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron . Prior Knowledge Questions (Do these BEFORE using the Gizmo.) 1. There are eight markers in a full set, but Flora and Frank each only have seven markers.

Student Exploration: Covalent Bonds

Gizmos Exploration Guide Covalent Bonding Single fluorine atoms are even more unstable than Mr. Snyder. In fact, single fluorine atoms are never found alone in nature. Fluorine atoms have a strong tendency to form bonds and will react strongly with other substances to form ionic or covalent compounds. An atom is most stable when it has a full outer energy level, or valence shell.

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Student Exploration: Covalent Bonds Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron Prior Knowledge Questions (Do these BEFORE using the Gizmo.) There are eight markers in a full set, but Flora and Frank each only have seven markers.

Student Exploration: Covalent Bonds (ANSWER KEY)

Covalent bonds. Covalent bonds form as valence electrons are shared between two atoms. Lewis Structures and structural formulas are common ways of showing the covalent bonding in organic molecules. Discuss, from a chemical viewpoint, why it is important to understand the three-dimensional geometry of a molecule.

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