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Atoms are the smallest unit of matter that cannot be divided using any chemical method. They do consist of smaller parts, but can only be broken by nuclear reactions. The three parts of an atom are protons, neutrons, and electrons. Protons carry a positive electrical charge.

Basic Atomic Structure and Atomic Theory - Study Guide

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Unit 3 Atomic Structure- Guided Notes The Periodic Table ... - Add you answers from number 1 together - Round your answer to 2 decimal places • Practice: 1. A sample of Cesium is 75% ^{133}Cs , 20% ^{132}Cs , and 5% ^{134}Cs . What is the average atomic mass? 2. Carbon-12 is 98% abundant and Carbon-14 is 2% abundant.

Name: Period: Unit 3 Atomic Structure- Guided Notes

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Mr. Christopherson / Atomic Structure

Procedure: On Monday/Tuesday, students will take notes on atomic structure in their binders, and complete a concept map, as described on the last page of the notes. Click here for the Atomic Structure Notes and Concept Map instructions. (Due W/R {11/9, 11/10}) Watch this video for an overview on atomic structure.

Atomic Structure Notes, Concept Map and Ion Guided Reading

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This guided inquiry lesson enables students to construct their own understanding of how to determine the number of protons, electrons, and neutrons in an atom. They also discover the role of protons, electrons, and neutrons in determining the structure, charge, and identity of an atom.

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Bill Gates Sr., a lawyer and the father of Microsoft's co-founder, who stepped in when appeals for charity began to overwhelm his billionaire son and started what became the world's largest ...

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